Water Sample Data Documentation

Introduction

During the two Arabesque cruises over 200 different parameters were measured on water samples by 38 principal investigators using a wide range of protocols. The aim of this document is to allow the protocol used to obtain any particular data value within the BOTDATA table to be determined with ease.

To help you find the information you require quickly, the document is subdivided into sections that describe groups of closely related parameters. These are listed below as a series of hot links. Each section starts with the definition of the parameter codes covered, followed by a list of who measured one or more of those parameters. Next, there is a protocol section describing the methods used by each principal investigator. Finally, there may be comments on data quality that have been noted by BODC or have come to our attention.

<TIP> If you want to find out a how a particular parameter was measured and know the parameter code then the fastest way to find the information you require is to use the *Acrobat* 'find' tool to search for the parameter code. Then use the 'find' tool again to search for the name of the principal investigator. This will take you straight to the protocol description you require.

Document Index

Bacterial Production, Abundance and Losses

Bacterial abundance, grazing losses and regeneration data, plus thymidine and leucine uptake data.

Dissolved Organic Carbon and Nitrogen

High temperature catalytic oxidation measurements.

Particulate Organic Carbon and Nitrogen

Those parameters loosely described as 'POC' and 'PON'

Nutrients

Nitrate plus nitrite, nitrite, phosphate, silicate, ammonia and urea data.

Amino Acids and Fatty Acids

Total free amino acids plus some seventy individual fatty acids.

Carbonate System Parameters

Dissolved total inorganic carbon and pCO2.

Pigments

Chlorophyll-a determined by a range of techniques, including data derived from calibrated *in situ* fluorometers, plus a full suite of pigments determined by HPLC.

Dimethylsulphide and its Precursors

Dimethylsulphide plus DMSP and DMSO.

Methane and Nitrous Oxide

Dissolved nitrous oxide and methane data.

Methylamines

Concentrations of dissolved monomethylamine, dimethylamine and trimethylamine.

Dissolved Tracers

Concentrations of sulphur hexafluoride (SF₆).

Dissolved Oxygen

Dissolved oxygen concentrations including data derived from calibrated *in situ* oxygen probe data.

Hydrography

Temperature, salinity, density and attenuance data that have largely been derived from CTD data together with calibration bottle salinity and reversing thermometer data.

Irradiance

Light meter data at bottle firing depths.

Microzooplankton Biomass and Grazing

Microzooplankton abundance, biomass and grazing together with data on heterotrophic and photosynthetic nanoflagellate abundances.

Phytoplankton Species Counts

Abundance of phytoplankton taxa.

AFC counts

Abundance of four broad taxonomic groupings as defined by Automated Flow Cytometry.

References

Full references for the papers cited in the protocol descriptions.

Bacterial Production Abundance and Losses

Parameter Code Definitions

TBCCMDPZ Total bacterial cell numbers per ml

Optical microscopy of DAPI stained samples

CBCCMDPZ Cyanobacteria cell numbers per ml

Optical microscopy

UPLERIP4 Leucine uptake rate pmol/litre/hour

Isotope doped incubation, 0.2µm Nuclepore filtered

UPTHRIP4 Thymidine uptake rate pmol/litre/hour

Isotope doped incubation, 0.2µm Nuclepore filtered

Originator Code Definitions

3 Dr. A. Pomroy Plymouth Marine Laboratory

105 Dr. T. Weisse Max Planck Institute

Originator Protocols

Dr. A. Pomroy

Samples were collected from 10 depths corresponding to 97%, 55%, 32.6%, 19.9%, 13.8%, 6.9%, 4.6%, 3%, 2.1% and 1% of surface irradiance, using pre-dawn CTD casts.

Tritiated thymidine incorporation experiments followed the methods of Fuhrman and Azam (1982) and the leucine incorporation experiments followed the methods of Simon and Azam (1989), modified to include the cold trichloroacetic acid (TCA) extraction method of Chin-Leo and Kirchman (1988). Five replicate, 10 ml aliquots from each depth sampled were transferred to sterile, polystyrene, tissue-culture tubes and placed in an incubator in the dark, at *in situ* temperatures and allowed to acclimatise for 15 minutes prior to the addition of the isotope. Electron microscope grade glutaraldehyde was added to one replicate sample from each depth at a final concentration of 2.5% by volume to act as controls. ³H-thymidine or ³H-leucine was added to each tube to give final concentrations of 5 and 10 nM respectively.

The samples were incubated for one to one and a half hours, but time-course assays showed that incorporation was linear for two hours and frequently longer.

At the end of the incubation, samples were transferred to an ice/water bath and ice-cold trichloroacetic acid (TCA) added to give a final concentration of 5% by volume. The samples were left in the water bath for 15-30 minutes and filtered through 25mm 0.2 micron pore-size, track-etched, polycarbonate membrane filters. Each filter was rinsed five times with 1ml 5% ice-cold TCA, placed in a scintillation vial and stored in a desiccator with active silica gel for 24 hours. At the end of this period, the samples were

counted in an LKB Rackbeta 1219 liquid scintillation counter. Counting efficiency was determined by an external standard, channels ratio method and checked by the occasional addition of internal standards.

Bacterial abundance

Samples were fixed with 2.5% by volume, 0.2 micron filtered, electron microscope grade glutaraldehyde, stained immediately with DAPI (4'6-diamidino-2-phenylindole) as described by Porter and Feig (1980) and filtered.

Samples were either examined immediately or stored frozen at -20 °C until being examined back at the laboratory. Fluorescent bacteria were counted with an epifluorescence microscope by the method of Hobbie et al. (1977). The microscope used was a Leitz Ortholux II equipped with a 50W HBO light source, Ploempak 2.2 fluorescence vertical illuminator with filter block A and an NPL Fluorotar 100/1.32 oil objective lens.

Dr. T. Weisse

Samples were collected from up to 10 depths on pre-dawn CTD casts and mixed to obtain one integrated sample of the euphotic zone. The nominal depth given refers to the shallowest depth sampled.

Up to five 300ml aliquots of the mixed water were poured into clean 500ml polycarbonate Erlenmeyer bottles, pre-filtered through a 100 micron screen to remove larger predators, inoculated and dark incubated at ambient temperatures for 1 hour.

The uptake of tritiated leucine labelled bacteria was measured by labelling with (4,5-³H)Leucine of high specific activity (171 Ci/mmol) for 12 - 25 hours. After labelling, the bacteria were heat killed at a temperature of 70-80 °C and filtered onto 0.2 micron Nuclepore filters and resuspended in sterile filtered sea water. This suspension was added to the Erlenmeyer bottles in a ratio of 1:4 to 1:6 to the mixed water samples, to which Penicillin/Streptomycin antibiotics had been added. A control bottle was fixed with 1% final concentration buffered glutaraldehyde. After incubation for 1 hour, the samples were killed by adding 2ml of 5% ice-cold trichloroacetic acid (TCA) and rinsing 4 times with 2ml aliquots of 5% ice-cold TCA after sequential filtering onto 8, 1 and 0.2 micron Nuclepore filters. The filters were stored dry in 5ml scintillation vials for 24 hours. At the end of this period, 2.5ml OptiPhase 'HisSafe' scintillation cocktail was added and the samples were counted in an LKB Rackbeta 1219 liquid scintillation counter. Counting efficiency was determined by an external standard, channels ratio method and checked by the occasional addition of internal standards. The decline in bacteria cell numbers during the period of incubation was used as an independent measurement of total bacterial loss rates.

Dissolved Organic Carbon

Parameter Code Definitions

CORGCOD1 Dissolved organic carbon (µmoles/litre)

High temperature platinum catalytic oxidation (GF/F filtered)

SEOCCOD1 Dissolved organic carbon standard error (µmoles/litre)

High temperature platinum catalytic oxidation (GF/F filtered)

Originator Code Definitions

13 Dr A.E.J.Miller Plymouth Marine Laboratory

Originator Protocols

Samples were taken from the CTD rosette and filtered through GF/F filters. Ultra-clean handling techniques were used throughout.

The analytical technique involves the direct injection of acidified and decarbonated sea water onto a platinised alumina catalyst at high temperature (680-900 °C) under an atmosphere of oxygen or high purity air. Quantitative production of CO₂ gas allows DOC concentrations to be determined using a CO₂-specific infrared gas analyser (IRGA).

Analyses were undertaken at sea using a Shimadzu TOC-5000 HTCO analyser fitted with a LiCor Li6252 IRGA. This overcame the problems associated with using the standard TOC-5000 IRGA on an unstable platform.

Great care was taken to quantify blank signals generated at all stages of the analytical procedure and to correct the data for them.

A more detailed description of the protocols followed may be found in Miller et al (1993).

For total dissolved nitrogen (TDN), the analytical technique involves the direct injection of acidified and decarbonated sea water onto a platinised alumina catalyst at high temperature (680-900 °C) under an atmosphere of oxygen or high purity air. Quantitative production of the nitric oxide radical allows total dissolved nitrogen concentrations to be determined using a nitrogen-specific chemiluminescence detector.

Analyses were undertaken at sea using a Shimadzu TOC-5000 HTCO analyser fitted with an Antek 705-D chemiluminescence detector. The combustion products travelled through a Drierite trap (97% CaSO₄, 3% CoCl₃) and a membrane (permeation tube) drier to remove any trace of water. The dried nitric acid radical was then reacted with ozone to produce the excited chemiluminescent nitrogen species and passed to the detector. Each sample was injected four times with each injection cycle taking 5.5 minutes.

POC and PON

Parameter Code Definitions

CORGCZP1 Particulate organic carbon (acidified)

Acid fumed then C/N analyser (GF/F filtered)

Micromoles/litre

NTOTCNP1 Particulate total nitrogen ("PON")

Carbon/nitrogen analyser (GF/F filtered)

Micromoles/litre

Originator Code Definitions

83 Dr. Tim Fileman, Plymouth Marine Laboratory

Originator Protocols

Dr, Tim Fileman

SAP collection

Challenger Oceanics *in situ* stand-alone pumps (SAPs) were used to sample particulate material. The instruments are operated by a programmable timer to ensure that the pump only operates when in position at the desired depth. Membrane filters with a 0.4 micron pore size were used to collect the particulate material. On recovery the filters were rinsed and dried in clean conditions.

Replicate 500 ml aliquots were taken from CTD rosette bottles or the underway non-toxic sea water supply. After an initial screening through a 200 micron mesh, to prevent spurious results caused by large zooplankton, the samples were filtered through 25mm GF/F filters. Additional aliquots were taken on some stations and filtered through 30 micron pore filters to give additional data for the >30 micron size fraction. Samples were frozen at -20 °C until analysed back at the laboratory.

The samples were acidified with sulphur dioxide to remove carbonates and then dried at 50 °C for 2 days. The samples were then encapsulated in squares of pre-combusted aluminium foil in a 4.5mm press.

The samples were analysed in a Carlo Erba NA1500 elemental analyser at a reactor temperature of 1030 °C and a helium carrier flow rate of 120 ml per minute. Calibration was effected with standards of acetanilide assayed on a calibrated Cahn 25 balance. Filter and sea water blanks were analysed and used to correct the data.

Nutrients

Parameter Code Definitions

NTRZAATX Nitrate + nitrite concentration µmol/litre

Colorometric autoanalysis (unfiltered)

NTRIAATX Nitrite concentration µmol/litre

Colorometric autoanalysis (unfiltered)

AMONAATX Ammonium concentration µmol/litre

Colorometric autoanalysis (unfiltered)

PHOSAATX Phosphate concentration µmol/litre

Colorometric autoanalysis (unfiltered)

SLCAAATX Silicate concentration µmol/litre

Colorometric autoanalysis (unfiltered)

UREAAATX Urea concentration µmol/litre

Colorometric autoanalysis (unfiltered)

Originator Code Definitions

62 Dr. E.M.S. Woodward Plymouth Marine Laboratory

Originator Protocols

Standard autoanalyser methods were used as described in Rees *et al* (1995) for the measurement of nitrate+nitrite, nitrite, phosphate, silicate and urea. A new semi-continuous fluorescence analytical technique was used for ammonia, capable of nanomolar detection levels. In oligotrophic water, a nanomolar chemiluminescent analysis system was used.

Carbonate System Parameters

Parameter Code Definitions

TCO2CBTX Total dissolved inorganic carbon (TCO₂) µmol/litre

Coulometric analysis (unfiltered)

PCO2GCO1 pCO₂ (parts per million)

Gas chromatography; shower-head equilibrator

Originator Code Definitions

57 Dr C. Robinson Plymouth Marine Laboratory
31 Dr. S. Knox Plymouth Marine Laboratory

Originator Protocols

Dissolved inorganic carbon was measured coulometrically. The instrument used was a Coulometric Incs model 5011 coulometer as described in Robinson and Williams (1991). The mean analytical precision was estimated as 0.5 - 1 micromole/kg.

Carbon dioxide (pCO2) was measured using a 'shower head' type equilibrator from which the gas phase was sampled. The equilibrated gas was carried in hydrogen and passed over a catalyst which converts the carbon dioxide to methane. The gas stream was then passed to a flame ionisation detector (FID), Hewlett-Packard 5890A- GC which uses compressed air as the combustion make-up gas. Primary calibration was by reference to a volumetrically prepared standard (pure carbon dioxide in zero air). All data are given at *in situ* temperature and corrected for atmospheric pressure changes, in microatmospheres. Precision studies have shown a standard deviation of no more than 1.5 microatmospheres for this method; repeated measurements of standards show a typical coefficient of variation to be no more than 0.2%.

Pigments

Parameter Code Definitions

CPHLFLP1 Fluorometric chlorophyll-a µg/litre

Fluorometric assay of acetone extract (GF/F filtered)

CPHLHPP1 HPLC chlorophyll a µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

CPHLPR01 Calibrated CTD chlorophyll µg/litre

Aquatrakka fluorometer calibrated against HPLC samples

ALLOHPP1 Alloxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

BUTAHPP1 Butanoyloxyfucoxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

C1C2HPP1 Chlorophyll c1c2 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

CHLBHPP1 Chlorophyll b µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

CLC3HPP1 Chlorophyll c3 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

DIADHPP1 Diadinoxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

DVCAHPP1 Diavinyl chlorophyll a µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

FUCXHPP1 Fucoxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

HEXOHPP1 Hexanoyloxyfucoxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PBA1HPP1 Phaeophorbide-a1 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PBA2HPP1 Phaeophorbide-a2 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PBA3HPP1 Phaeophorbide-a3 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PBA4HPP1 Phaeophorbide-a4 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PERIHPP1 Peridinin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PTA1HPP1 Phaeophytin-a1 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

PTA2HPP1 Phaeophytin-a2 µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

ZEOXHPP1 Zeaxanthin µg/litre

Reverse-phase HPLC assay of acetone extract (GF/F filtered)

Originator Code Definitions

5 Dr R. Barlow Plymouth Marine Laboratory

Originator Protocols

Samples of one to two litres were taken from all shallow biogeochemistry CTD casts, filtered onto GF/F filters and immediately frozen in liquid nitrogen. The filters were extracted into 90% acetone, an aliquot was taken and injected onto a C-8 reverse phase column for high pressure liquid chromatographic (HPLC) separation and quantification of some 20 chlorophyll and carotenoid pigments, using both absorbance at 440nm and fluorescence (excitation at 405nm, emission at 670nm) detection.

Dimethylsulphide and its Precursors

Parameter Code Definitions

DMSOGCD4 Dissolved dimethylsulphoxide nanomoles/litre

Gas chromatography on DMS released by enzyme digestion

(Millipore prefiltered)

DMSPGCD4 Dissolved dimethylsulphoniopropionate nanomoles/litre

Gas chromatography (Millipore prefiltered)

DMSXGCD4 Dissolved dimethylsulphide nanomoles/litre

Gas chromatography (Millipore prefiltered)

Originator Codes

68 Dr. A. Hatton Univ. of East Anglia

Originator Protocols

Water samples were collected from the continuous pumped sea water supply or from 10 litre Go-Flo bottles attached to the CTD rosette, using 500 ml ground glass bottles sealed with ground glass stoppers to leave minimal head space. The water was supplied to the bottom of the bottle using a silicate tube and the water allowed to overflow to prevent bubble entrapment.

Samples were transferred to the ship's laboratory for immediate DMS analysis by purge and trap gas chromatography. A volumetric aliquot of water was injected into the purging vessel through a Millipore pad pre-filter. Trace gases were extracted from the water by a stream of high grade nitrogen which subsequently passed through a cold trap (-150°C) where the gases were concentrated before being heated and injected into the gas chromatograph.

The gases were analysed using a Chromosil 330 column, isothermal at 40°C with flame photometric detection (FPD). Full details are given in Turner *et al* (1990). All DMS analyses were completed on board ship. The instrument was calibrated at the start and end of each sampling run using a stock DMS standard.

An aliquot of the purged water sample was treated with 10M NaOH (to decompose DMSP to DMS). The dissolved DMSP was quantified as DMS as described above. The filter pad was also treated with 10M NaOH and the DMS released was quantified to give particulate DMSP.

A second aliquot of purged water was treated with a solution containing purified DMSO reductase, ethylenediaminetetraacetic acid (EDTA) and flavin mononucleotide (FMN) to convert DMSO into DMS. Dissolved DMSO was then determined as DMS as described above. Further details of this technique may be found in Hatton *et al* (1994).

Methane and Nitrous Oxide

Parameter Code Definitions

CH4DGCDX Dissolved methane nanomoles/litre

Single phase equilibration gas chromatography

CH4SGCDX Dissolved methane saturation (%)

Calculated from methane concentration

DN2OGCTX Dissolved nitrous oxide (N2O) nanomoles/litre

Single phase equilibration gas chromatography

SN2OGCTX Dissolved nitrous oxide saturation (%)

Calculated from nitrous oxide concentration

Originator Codes

22 Prof. N.J.P. Owens Univ. of Newcastle

Originator Protocols

The dissolved seawater concentrations of nitrous oxide and methane (both important greenhouse gases) was determined simultaneously to high precision using a fully automated headspace equilibration gas chromatographic technique.

Samples were collected in 1 litre glass volumetric flasks and allowed to overflow by at least three volumes. Immediately following collection, the samples were stabilised by the addition of 200µl 0.25M aqueous mercuric chloride and the flasks were stoppered and inverted to disperse the HgCl₂. Analysis followed immediately; the samples were thermally equilibrated to 25°C in a water bath and then equilibrated with a headspace of known nitrous oxide and methane. The equilibrated gases were dried and passed over a Carbosorb column to remove CO₂ before being separated on Porapak Q columns and injected into the carrier gas lines (ultra high purity nitrogen) of two Shimadzu GC 8A gas chromatographs. The first was equipped with a flame ionisation detector (FID) for the analysis of methane; the second with an electron capture detector (ECD) for the measurement of nitrous oxide.

Two high mixing ratio primary standards were used for calibration, containing 10ppmv nitrous oxide, 5.2 ppmv methane and 20ppmv nitrous oxide, 8.5ppmv methane respectively. For routine calibrations, four secondary standards were prepared from these primary standards by pressure dilution (Upstill-Goddard *et.al.* 1990); estimates for their accuracy are 1.5% for methane and 2% for nitrous oxide. Analytical precisions are better than 0.5% for nitrous oxide and 0.4% for methane.

Methylamines

Parameter Code Definitions

DIMAFITX Dissolved dimethylamine concentration nanomoles/litre

Flow injection gas diffusion (unfiltered)

MOMAFITX Dissolved monomethylamine concentration nanomoles/litre

Flow injection gas diffusion (unfiltered)

TRMAFITX Dissolved trimethylamine concentration nanomoles/litre

Flow injection gas diffusion (unfiltered)

Originator Code Definitions

71 Dr. S.W. Gibb Plymouth Marine Laboratory

Originator Protocols

Water samples were collected from the CTD casts using either 250 ml gas-tight polythene bottles or 100 ml glass syringes. The methylamine concentrations were determined on board using Flow Injection Gas Diffusion coupled to Ion Chromatography (FIGD-IC).

The FIGD-IC procedure is described in Gibb *et al* (1995). This is a novel technique that allows the simultaneous measurement of methylamines and ammonia at nanomolar levels. Briefly, the ammonia and methylamines were deprotonated to their free, volatile forms through alkali admixing (NaOH, pH>12) and selectively transferred by diffusion across a gas-permeable membrane into a dynamic, acidic acceptor stream in which they were enriched in their cationic forms. Chelation of the alkali earth metals in the samples with EDTA, under thermodynamically optimised conditions, was used to prevent the precipitation of their hydroxides under the elevated pH conditions. The enriched acceptor stream was then transferred directly into an ion chromatograph in which the NH₄⁺ and methylamines were separated within 15 minutes in an acidic eluent and quantified by chemically suppressed conductimetric detection.

Dissolved Sulphur Hexafluoride

Parameter Code Definitions

DSF6GCDX Dissolved sulphur hexafluoride (SF₆) femtomoles/litre

Gas chromatography- electron capture detection (unfiltered)

Originator Code Definitions

106 Dr. C. Law Plymouth Marine Laboratory

Originator Protocols

Water samples were collected from the CTD casts using syringes. The samples were then injected into the sparge tower of the gas chromatograph under a vacuum (500mm Hg) and mixed with sparge gas (oxygen-free nitrogen). Degassing of SF $_6$ from the sample was accelerated by the drawing in and rapid expulsion of the sample through 0.5mm orifices. The sparged SF $_6$ was then passed through magnesium perchlorate followed by a Nafion drier to absorb any water vapour and then directed through a cryogenically cooled Porapak Q trap (Upstill-Goddard et.al. 1991) to be stripped. The trap containing the sparged SF $_6$ was then isolated and raised out of the propanol, being heated from -70°C to 80°C in 25 seconds and the sparged SF $_6$ was transported to the chromatographic column where it was eluted after 50 seconds and passed to the electron capture detector. Oxygen and similar eluents were retained by molecular sieve columns.

Calibration was by reference to standards prepared by pressure dilution (Upstill-Goddard et.al., 1991) of 0.1% SF₆ in nitrogen mixture (Spantech Products Ltd., UK) in 9.4 litre bottles. Calibration precision for each GC unit showed a mean standard deviation of 2.5% or better.

Dissolved Oxygen

Parameter Code Definitions

DOXYPR01 Calibrated CTD Beckmann oxygen probe micromoles/litre

Beckmann probe calibrated against Winkler samples

DOXYWITX Dissolved oxygen concentration micromoles/litre

Winkler titration

OXYSBB01 Dissolved oxygen saturation (%)

Benson & Krause algorithm from Beckmann probe data

Originator Codes

27 Dr. J. Dickson Univ. of Plymouth

Originator Protocols

Dr. J. Dickson

The technique used was the standard automated Winkler titration, as described in Williams and Jenkinson (1982).

British Oceanographic Data Centre

The oxygen probe concentrations and saturations in table BOTDATA have been obtained by extracting the data from the CTD **downcasts** corresponding to the bottle firing depths. The CTD system used was a Neil Brown Mk3B with a non-pulsed membrane Beckmann oxygen sensor. Oxygen data were calibrated against Jo Dickson's water bottle data.

Oxygen saturations were computed using calibrated temperature and salinity data and the algorithm of Benson and Krause (1984).

Hydrography

Parameter Code Definitions

ATTNZR01 Red light attenuance (per metre)

660nm transmissometer

POTMCV01 Potential temperature degrees Celsius

Computed from CTD data using UNESCO function POTEMP

PSALBSTX Bench salinometer salinity (PSU)

Guildline Autolab salinometer

PSALST01 CTD salinity (PSU)

Derived from CTD conductivity and temperature measurements

SIGTPR01 CTD sigma-theta

Computed using UNESCO function SVAN (stearic volume

anomaly)

TEMPRTNX Reversing thermometer temperature (degrees Celsius)

Digital SIS reversing thermometers, mounted on CTD bottle

TEMPST01 CTD temperature (degrees Celsius)

CTD platinum resistance thermometer

TOKGPR01 Micromolar to micromoles/kg conversion factor

Calculated from CTD data

Originator Code Definitions

1 W. Miller Research Vessel Services1 J. Wynar Research Vessel Services

Originator Protocols

In most cases where the parameter code ends in '01', the values have been obtained by BODC software which extracts CTD **downcast** data corresponding to the bottle firing depths. This ensures an internally consistent data set across all cruises regardless of whether or not the upcast data were made available. The method is prone to errors if significant changes occur to water column structure during the cast. In all cases, further details about the CTD data may be obtained from the CTD document for the relevant cruise.

The conversion factor TOKGPR01 is 1000/(1000+sigma-theta) and is stored to allow sample data stored in concentration per litre to be converted to concentration per kilogram.

An RVS Neil Brown Mk 3B CTD was used with a SeaTech 25cm path length red light (661 nm) transmissometer fitted to the cage. Temperatures were checked against SIS digital reversing thermometers and salinity calibrated against bottle salinity data analysed on board on a Guildline Autolab salinometer. Transmissometer data were corrected for light source decay using the air readings during the cruise and the air readings quoted by the manufacturer (SeaTech).

Irradiance

Parameter Code Definitions

IRRDPP01 Downwelling 2-pi PAR irradiance (microEinsteins/square

metre/second)

Hemispherical photodiode light meter mounted on CTD frame

IRRUPP01 Upwelling 2-pi PAR irradiance (microEinsteins/square

metre/second)

Hemispherical photodiode light meter mounted on CTD frame

Originator Code Definitions

16 BODC

Originator Protocols

The data presented in the BOTDATA table are derived from CTD **downcast** data at the bottle firing depths. Note that the interpolation was done on log transformed data to allow a linear technique to be used.

The data were collected by Plymouth Marine Laboratory designed light meters based on a photodiode under a hemispherical translucent white plastic cap. The sensors were designed to collect light across the visible portion of the spectrum.

The light meters were fitted to the CTD frame with the downwelling instrument projecting above the top of the bottle rosette and the upwelling instrument attached to the base of the cage. This gave a physical separation of approximately two metres.

The data were logged as voltages and converted to W/m² using laboratory calibrations. The calibrations used were over five years old. The data were converted to $\mu E/m^2/s$ using an empirically derived conversion factor of 3.75.

Microzooplankton Biomass and Grazing

Parameter Code Definitions

MZBCMITX Microzooplankton biomass (mg carbon/cubic metre)

Calculated from cell numbers determined by optical microscopy

MZBNMITX Microzooplankton abundance (cells/ml)

Optical microscopy

P400E00A Autotrophic nanoflagellates (2-20 µm) per mil

Epifluorescence microscopy with DAPI/proflavine stain

P400E00B Heterotrophic nanoflagellates (2-20 µm) per mil

Epifluorescence microscopy with DAPI/proflavine stain

Originator Code Definitions

84 Dr. P.H. Burkill Plymouth Marine Laboratory 41 E.S. Edwards Plymouth Marine Laboratory

Originator Protocols

Water samples were obtained from water bottles deployed on a CTD rosette. These were fixed with 1% Lugol's iodine and the microzooplankton were counted using an image analysis system coupled to an inverted microscope. Fixed samples were gently mixed and sub-samples of 30-100 ml were concentrated overnight in sedimentation chambers. Each sample was examined at a magnification of x300 and all grazers > circa 10 microns were counted. Cells were identified to genus level whenever possible.

Live video work together with fluorescence microscopy at sea enabled the separation of heterotrophic dinoflagellates from phototrophic forms. All ciliates were assumed to be heterotrophic. In order to obtain a more accurate identification of some ciliates, Protargol silver staining was carried out on a number of samples.

The biomass was determined using methods detailed in JGOFS protocols (Burkill et al, 1994). The image analysis system was used to generate data on the surface area of each cell. These were converted to cell volume using geometric formulae and standard volume to carbon conversion factors were applied for different taxa. Individual cell carbon volumes were integrated for discrete taxa to determine the biomass of those taxa in each water sample.

Samples were also collected for the determination of nanoplankton (2-20 micron) abundance and biomass. Samples were fixed in 0.3% glutaraldehyde, dual stained with DAPI and proflavine and filtered onto 0.4 micron black polycarbonate filters. Cells were counted by epifluorescence microscopy. Heterotrophs were distinguished from autotrophs by the presence or absence of chlorophyll autofluorescence. 1-200 flagellate cells were counted per filter and cell dimensions were measured with an ocular micrometer.

Flagellate cell volumes were calculated assuming they were ellipsoids. Biovolumes were converted to biomass using appropriate carbon conversion factors.

Natural microbial populations were incubated either *in situ* or on board ship using the dilution technique described by Landry and Hassett (1982). Time course experiments were run under different dilutions and the specific growth of phytoplankton determined. Water samples were collected at dawn from a depth of 10m using 30 litre Niskin bottles. Half of this water was filtered using a Gelman 0.2 micron mini capsule filter. A known volume of this 'predator and prey free' water was added to polycarbonate bottles. Each bottle was gently topped up with 200 micron screened, unfiltered water generating triplicate dilutions of 100%, 70%, 40% and 10%. Incubation was carried out over 24 hours. Subsamples were taken from each bottle at T₀ and T₂₄ for determination of chlorophyll and fixation in Lugol's iodine for estimation of microzooplankton abundance. Chlorophyll was determined by extraction of 90% acetone, using a highly sensitive fluorometer. Phytoplankton mortality due to grazing was determined from alteration in the specific growth rate.

Qualitative analysis of microzooplankton herbivory was determined by incubating natural microbial populations with different types of fluorescently labelled algae (FLA). Cultured algae were heat stained with a fluorochrome, 5-(4,6-dichlorotriazan-2-yl) aminofluorescein (DTAF). Water samples were inoculated with FLA stock and experiments run for 60 minutes. Subsamples were collected over the experimental period and fixed in 1% Lugol's iodine. Up to 110 ml of sample were settled for 24 hours prior to analysis and the Lugol's coloration cleared with sodium thiosulphate. The FLAs present in each microzooplankton cell were counted and FLA uptake rates were calculated from the change in the average number of FLAs per individual with time. The mean cellular uptake rates were calculated for those taxa taking up FLAs in each experiment.

Phytoplankton

Parameter Code Definitions

P002M00Z	Actinocyclus spp. Optical microscopy	per mil
P012M00Z	Asteromphalus spp. Optical microscopy	per mil
P012M02Z	Asteromphalus sarcophagus Optical microscopy	per mil
P018M04Z	Bacteriastrum furcatum Optical microscopy	per mil
P018M05Z	Bacteriastrum elongatum Optical microscopy	per mil
P018M06Z	Bacteriastrum solitarium Optical microscopy	per mil
P021M06Z	Biddulphia mobiliensis Optical microscopy	per mil
P028M01A	Cerataulina pelagica (30µm) Optical microscopy	per mil
P028M01Z	Cerataulina pelagica Optical microscopy	per mil
P030M00C	Chaetoceros resting spores Optical microscopy	per mil
P030M00Z	Chaetoceros spp. Optical microscopy	per mil
P030M01Z	Chaetoceros affine Optical microscopy	per mil
P030M02A	Chaetoceros atlanticum v. neapol. Optical microscopy	per mil
P030M04Z	Chaetoceros boreale Optical microscopy	per mil
P030M05Z	Chaetoceros breve Optical microscopy	per mil

P030M09Z	Chaetoceros compressum Optical microscopy	per mil
P030M14Z	Chaetoceros costatum Optical microscopy	per mil
P030M19Z	Chaetoceros decipiens Optical microscopy	per mil
P030M20Z	Chaetoceros densum Optical microscopy	per mil
P030M21Z	Chaetoceros didymum Optical microscopy	per mil
P030M36Z	Chaetoceros laciniosum Optical microscopy	per mil
P030M39Z	Chaetoceros messanense Optical microscopy	per mil
P030M43Z	Chaetoceros peruvianum Optical microscopy	per mil
P030M70Z	Chaetoceros anastomosans Optical microscopy	per mil
P030M72Z	Chaetoceros diversus Optical microscopy	per mil
P030M73Z	Chaetoceros dadayii Optical microscopy	per mil
P030M74Z	Chaetoceros debilis Optical microscopy	per mil
P030M75Z	Chaetoceros lauderi Optical microscopy	per mil
P030M77Z	Chaetoceros similis Optical microscopy	per mil
P030M80Z	Chaetoceros saltans Optical microscopy	per mil
P033M01A	Corethron criophillum (15µm) Optical microscopy	per mil
P033M01Z	Corethron criophillum Optical microscopy	per mil

P034M01Z	Coscinodiscus africanus Optical microscopy	per mil
P034M22Z	Coscinodiscus oculis-iridis Optical microscopy	per mil
P034M28Z	Coscinodiscus thorii Optical microscopy	per mil
P040M02Z	Detonula pumila Optical microscopy	per mil
P048M01Z	Eucampia zoodiacus Optical microscopy	per mil
P048M02Z	Eucampia cornuta Optical microscopy	per mil
P052M00Z	Fragilaria spp. Optical microscopy	per mil
P058M00Z	Guinardia spp. Optical microscopy	per mil
P061M01Z	Hemiaulus hauckii Optical microscopy	per mil
P062M01Z	Hemidiscus cuneiformis Optical microscopy	per mil
P067M02Z	Lauderia annulata Optical microscopy	per mil
P068M01Z	Leptocylindrus danicus Optical microscopy	per mil
P068M02Z	Leptocylindrus mediterranea Optical microscopy	per mil
P072M00A	Pennates(small) Optical microscopy	per mil
P072M00C	Pennate (50µm) Optical microscopy	per mil
P073M10Z	Navicula planamembranacea Optical microscopy	per mil
P074M00A	Nitzschia spp. (70µm) Optical microscopy	per mil

P074M07Z	Nitzschia bicapitata Optical microscopy	per mil
P074M14Z	Nitzschia closterium Optical microscopy	per mil
P074M18Z	Nitzschia delicatissima Optical microscopy	per mil
P074M61Z	Nitzschia seriata Optical microscopy	per mil
P081M01Z	Planktoniella sol Optical microscopy	per mil
P084M00Z	Pleurosigma spp. Optical microscopy	per mil
P084M10Z	Pleurosigma directa Optical microscopy	per mil
P084M25Z	Pleurosigma planktonicum Optical microscopy	per mil
P087M00Z	Porosira spp. Optical microscopy	per mil
P087M02Z	Porosira dentuculata Optical microscopy	per mil
P093M02E	Rhizosolenia alata (15µm) Optical microscopy	per mil
P093M02F	Rhizosolenia alata (2µm) Optical microscopy	per mil
P093M02G	Rhizosolenia alata (5µm) Optical microscopy	per mil
P093M02Z	Rhizosolenia alata Optical microscopy	per mil
P093M06Z	Rhizosolenia bergonii Optical microscopy	per mil
P093M09Z	Rhizosolenia cylindrus Optical microscopy	per mil
P093M10Z	Rhizosolenia calcar-avis Optical microscopy	per mil

P093M11Z	Rhizosolenia castracanei Optical microscopy	per mil
P093M13Z	Rhizosolenia fragilissima Optical microscopy	per mil
P093M14C	Rhizosolenia hebetata semispina Optical microscopy	per mil
P093M20Z	Rhizosolenia robusta Optical microscopy	per mil
P093M21Z	Rhizosolenia setigera Optical microscopy	per mil
P093M22A	Rhizosolenia shrubsolei (10µm) Optical microscopy	per mil
P093M22B	Rhizosolenia shrubsolei (5µm) Optical microscopy	per mil
P093M22Z	Rhizosolenia shrubsolei Optical microscopy	per mil
P093M23A	Rhizosolenia stolterfothii (large) Optical microscopy	per mil
P093M23B	Rhizosolenia stolterfothii (small) Optical microscopy	per mil
P093M23Z	Rhizosolenia stolterfothii Optical microscopy	per mil
P093M24Z	Rhizosolenia styliformis Optical microscopy	per mil
P096M01Z	Roperia tessellata Optical microscopy	per mil
P101M01Z	Skeletonema costatum Optical microscopy	per mil
P102M07Z	Stauroneis membranacea Optical microscopy	per mil
P110M01Z	Thalassionema nitzschiodes Optical microscopy	per mil
P110M02Z	Thalassionema bacillaris Optical microscopy	per mil

P111M00A	Thalassiosira spp. (10μm) Optical microscopy	per mil
P111M00B	Thalassiosira spp. (20μm) Optical microscopy	per mil
P111M00C	Thalassiosira spp. (40μm) Optical microscopy	per mil
P111M00E	Thalassiosira spp. (4µm) Optical microscopy	per mil
P111M00F	Thalassiosira spp. (30μm) Optical microscopy	per mil
P111M00G	Thalassiosira spp. (45µm) Optical microscopy	per mil
P111M00H	Thalassiosira spp. (120µm) Optical microscopy	per mil
P111M00I	Thalassiosira spp. (2µm) Optical microscopy	per mil
P112M01Z	Thalassiothrix frauenfeldii Optical microscopy	per mil
P112M03Z	Thalassiothrix delicatula Optical microscopy	per mil
P200M00Z	Dinoflagellates Optical microscopy	per mil
P207M01Z	Amphisolenia bidentata Optical microscopy	per mil
P207M02Z	Amphisolenia globosa Optical microscopy	per mil
P213M04Z	Ceratium candelabrum Optical microscopy	per mil
P213M05Z	Ceratium extensum Optical microscopy	per mil
P213M08Z	Ceratium furca Optical microscopy	per mil
P213M09Z	Ceratium fusus Optical microscopy	per mil

P213M13Z	Ceratium horridum Optical microscopy	per mil
P213M18Z	Ceratium macroceros Optical microscopy	per mil
P213M26Z	Ceratium tripos Optical microscopy	per mil
P213M30Z	Ceratium boehmii Optical microscopy	per mil
P213M33Z	Ceratium praelongum Optical microscopy	per mil
P213M34Z	Ceratium teres Optical microscopy	per mil
P213M38Z	Ceratium ranipes Optical microscopy	per mil
P213M39Z	Ceratium schroederi Optical microscopy	per mil
P219M03Z	Dinophysis brevisulcus Optical microscopy	per mil
P219M16Z	Dinophysis doryphorum Optical microscopy	per mil
P219M18Z	Dinophysis favus Optical microscopy	per mil
P228M14Z	Gonyaulax milneri Optical microscopy	per mil
P228M17Z	Gonyaulax polygramma Optical microscopy	per mil
P228M19Z	Gonyaulax spinifera Optical microscopy	per mil
P229M00Z	Gymnodinium spp.(autotrophic) Optical microscopy	per mil
P229M50Z	Gymnodinium splendens Optical microscopy	per mil
P229M91Z	Gymnodinium A (autotrophic) Optical microscopy	per mil

P230M01Z	Gyrodinium aureolum Optical microscopy	per mil
P230M11Z	Gyrodinium falcatum Optical microscopy	per mil
P235M00Z	Heteraulacus spp. Optical microscopy	per mil
P257M04Z	Prorocentrum compressum Optical microscopy	per mil
P257M06Z	Prorocentrum gracile Optical microscopy	per mil
P257M07Z	Prorocentrum triestinum Optical microscopy	per mil
P257M09Z	Prorocentrum minimum Optical microscopy	per mil
P266M00Z	Scrippsiella spp. Optical microscopy	per mil
P315M00Z	Cochlodinium spp. Optical microscopy	per mil
P322M00Z	Diplopsalopsis spp. Optical microscopy	per mil
P329M00Z	Gymnodinium spp. (heterotrophic) Optical microscopy	per mil
P330M04Z	Gyrodinium britannicum Optical microscopy	per mil
P330M14Z	Gyrodinium fusiforme Optical microscopy	per mil
P330M15A	Gyrodinium glaucum (small) Optical microscopy	per mil
P330M15Z	Gyrodinium glaucum Optical microscopy	per mil
P339M00Z	Kofoidinium spp. Optical microscopy	per mil
P345M00A	Noctiluca spp. (juvenile) Optical microscopy	per mil

P345M00B	Noctiluca spp. (motile) Optical microscopy	per mil
P345M01Z	Noctiluca scintillans Optical microscopy	per mil
P349M00Z	Oxytoxum spp. Optical microscopy	per mil
P349M01Z	Oxytoxum scolopax Optical microscopy	per mil
P353M01Z	Podolampas bipes Optical microscopy	per mil
P353M02Z	Podolampas palmipes Optical microscopy	per mil
P356M00Z	Pronoctiluca spp. Optical microscopy	per mil
P358M15Z	Protoperidinium curtipes Optical microscopy	per mil
P358M42Z	Protoperidinium oceanicum Optical microscopy	per mil
P358M54Z	Protoperidinium steinii Optical microscopy	per mil
P358M55Z	Protoperidinium elegans Optical microscopy	per mil
P360M01Z	Ptychodiscus noctiluca Optical microscopy	per mil
P361M01Z	Pyrocystis lunula Optical microscopy	per mil
P361M03Z	Pyrocystis fusiformis Optical microscopy	per mil
P361M04Z	Pyrocystis noctiluca Optical microscopy	per mil
P366M00A	Peridinians (large) Optical microscopy	per mil
P366M00B	Peridinians (small) Optical microscopy	per mil

P370M00A	Torodinium spp. (small) Optical microscopy	per mil
P370M01Z	Torodinium robustum Optical microscopy	per mil
P371M00Z	Warnowia spp. Optical microscopy	per mil
P400M00A	Flagellate 2µm Optical microscopy	per mil
P400M00B	Flagellate 4µm Optical microscopy	per mil
P404M04Z	Cryptomonad Optical microscopy	per mil
P410M00A	Holococcolithophorid 10µm Optical microscopy	per mil
P410M02A	Holococcolithophorid (small) Optical microscopy	per mil
P411M01Z	Acanthoica quattrospina Optical microscopy	per mil
P412M01Z	Anoplosolenia brasiliensis Optical microscopy	per mil
P413M01Z	Anthosphaera robusta Optical microscopy	per mil
P415M02Z	Calciopappus rigidus Optical microscopy	per mil
P416M01Z	Calciosolenia murrayi Optical microscopy	per mil
P417M00Z	Calyptrosphaera spp. Optical microscopy	per mil
P427M00Z	Halosphaera spp. Optical microscopy	per mil
P428M01Z	Helicosphaera carteri Optical microscopy	per mil
P432M01Z	Michaelsarsia elegans Optical microscopy	per mil

P433M00Z	Ophiaster spp. Optical microscopy	per mil
P436M00Z	Phaeocystis spp. Optical microscopy	per mil
P439M00Z	Pterosperma spp. Optical microscopy	per mil
P445M00A	Syracosphaera spp. (10 μm) Optical microscopy	per mil
P445M00B	Syracosphaera spp. A (20 μm) Optical microscopy	per mil
P445M00C	Syracosphaera spp. B (20 μm) Optical microscopy	per mil
P445M06Z	Syracosphaera pulchra Optical microscopy	per mil
P448M01Z	Emiliana huxleyi Optical microscopy	per mil
P449M00Z	Gephyrocapsa spp. Optical microscopy	per mil
P449M01Z	Gephyrocapsa oceanica Optical microscopy	per mil
P452M02Z	Umbellosphaera irregularis Optical microscopy	per mil
P500M17Z	Ciliates Optical microscopy	per mil
P510M01Z	Bodonids Optical microscopy	per mil
P521M05B	Mesodinium spp. (medium) Optical microscopy	per mil
P521M05C	Mesodinium spp. (small) Optical microscopy	per mil
P981M00Z	Halopappus spp. Optical microscopy	per mil
P982M00Z	Erythopsis spp. Optical microscopy	per mil

P983M01Z	Umbillicosphaera sibogae Optical microscopy	per mil
P984M01Z	Trichodesmium thibauthii Optical microscopy	per mil
P985M00A	Strombidium spp. (large) Optical microscopy	per mil
P985M00B	Strombidium spp. (small) Optical microscopy	per mil
P985M00C	Strombidium spp. (medium) Optical microscopy	per mil
P986M00Z	Pyramimonas spp. Optical microscopy	per mil
P987M01Z	Pachyneis gerlachii Optical microscopy	per mil
P988M01Z	Ornithocercus quadratus Optical microscopy	per mil
P989M00Z	Oolithus spp. Optical microscopy	per mil
P990M01Z	Histoneis hyalina Optical microscopy	per mil
P991M01Z	Florisphaera profunda Optical microscopy	per mil
P992M01Z	Crenalithus sessilis Optical microscopy	per mil
P993M01Z	Climacodium frauenfeldii Optical microscopy	per mil
P994M01Z	Ceratocorys horrida Optical microscopy	per mil
P995M01Z	Calcidiscus leptoporus Optical microscopy	per mil
P996M00Z	Brachydinium spp. Optical microscopy	per mil
P997M00Z	Asterolampra spp. Optical microscopy	per mil

Originator Codes

77 Dr. D.S. Harbour Plymouth Marine Laboratory

Originator Protocols

Water samples were taken from bottles deployed on the CTD and preserved in Lugol's lodine. Back in the laboratory, sedimented samples were examined by optical microscopy and the dominant species in the >5 micron size fraction were quantified.

Automated Flow Cytometry

Parameter Code Definitions

CBCCAFTX Cyanobacteria cell numbers per mil

Automated flow cytometry

PYEUAFTX Eukaryotic cell numbers per mil

Automated flow cytometry

PYPKAFTX Prokaryotic cell numbers per mil

Automated flow cytometry

PYTTAFTX Total cell numbers per mil

Automated flow cytometry

Originator Code Definitions

84 Dr. P.H. Burkill Plymouth Marine Laboratory 104 Dr. G.Tarran Plymouth Marine Laboratory

Originator Protocols

Samples were collected from all available depths of the shallow biogeochemistry CTD casts and a 400 microlitre aliquot was injected into a Becton Dickinson FACSort cytometer. Sensitivity was sufficient to determine cellular light scatter and fluorescences from prochlorophytes (0.6 micron size, approximate chlorophyll a content 1 femtogram) in surface waters. Protocols based on light scatter and fluorescence were used to characterise and quantify total phytoplankton concentrations and those of individual taxa (prochlorophytes, cyanobacteria and picoeukaryotes).

References

Armstrong, F.A.J., Stearns, C.R. and Strickland, J.D.H., 1967. The measurement of upwelling and subsequent biological processes by means of the Technicon Autoanalyser and associated equipment. *Deep Sea Res.* 14,381-389.

Barlow, R.G., Mantoura, R.F.C., Gough, M.A. and Fileman, T.W., 1993a.Pigment signatures of the phytoplankton composition in the north-east Atlantic during the 1990 spring bloom. *Deep Sea Res. II*, 40, 459-477.

Barlow, R.G., Mantoura, R.F.C., Gough, M.A. and Fileman, T.W., 1993b. Phaeopigment distribution during the 1990 spring bloom in the north-east Atlantic. *Deep Sea Res. I*, 40, 2229-2242.

Barlow, R.G., Cummings, D.G., Mantoura, R.F.C. and Fileman, T.W., 1996. Pigment chemotaxonomic distributions of phytoplankton during summer in the western Mediterranean. *Deep Sea Res. II*, in press.

Benson, B.B., Krause D. 1984. The concentration and isotopic fractionation of oxygen dissolved in fresh water and sea water in equilibrium with the atmosphere. *Limnol.Oceanogr.*, 29, 620-632.

Burkill, P.H., Edwards, E.S., Landry, M., Paranjape, M., Reckermann, M., Sieracki, Sleigh, M.A., Stoecker, D.K. and Verity, P., 1994. Microzooplankton biomass. In: Protocols for the Joint Global Ocean Flux Study (JGOFS) Core Measurements. Ed: A.H. Knap. Intergovernmental Oceanographic Commission Manual & Guides 29, 147-151. UNESCO, Paris.

Cauwet, G. 1994. HTCO method for dissolved organic carbon analysis in seawater: influence of catalyst on blank estimation, *Marine Chemistry*, 47, 55-64.

Chin-Leo, G. and Kirchman, D.L. 1988. Estimating bacterial production in marine waters from the simultaneous incorporation of thymidine and leucine. *Appl. Environ. Microbiol.*, 54, 1934-1939.

Dickson, A.G. 1990. Thermodynamics of the dissociation of boric acid in synthetic seawater from 273.15 to 298.15 K. *Deep Sea Research* 37, 755-766.

Dickson, A.G., 1993. PH buffers for sea-water media based on the total hydrogen-ion concentration scale. *Deep-Sea Research*, 40, 107-118.

Eberlein, K. and Kattner, G. 1987. Automatic method for the determination of orthophosphate and total dissolved phosphorus in the marine environment. *Fresenius Z. anal. Chem.*, 326, 354-357.

Fuhrman, J.A. and Azam, F., 1982. Thymidine incorporation as a measure of heterotrophic bacterioplankton production in marine surface waters: evaluation and field results. *Marine Biology*, 66, 109-120.

Gibb, S.W., Mantoura, R.F.C. and Liss, P.S., 1995. Analysis of ammonia and methylamines in natural waters by flow injection coupled to ion chromatography. *Analytica Chimica Acta*, 316, 291-304.

Goyet, C., Poisson, A., 1989. New determination of carbonic-acid dissociation constants in sea-water as a function of temperature and salinity. *Deep-Sea Research* 36, 163-165.

Grasshoff, K., Ehrhardt, M. and Kremling, K. eds. 1983. Methods of seawater analysis. *Verlag Chemie*.

Hansson, I., 1973. A new set of acidity constants for carbonic acid and boric acid in sea water. *Deep Sea-Research* 20, 461-478.

Hatton, A.D., Malin, G., McEwan, A.G. and Liss, P.S., 1994. *Analytical Chemistry*, 66, 4093-4096.

Hobbie, J.E., Daley, R.J. and Jasper, S., 1977. Use of Nuclepore filters for counting bacteria by epifluorescence microscopy. *Applied and Environmental Microbiology* 33, 1225-1228.

Holm-Hansen, O., Lorenzen, C.J., Holmes, R.W. and Strickland, J.D.H., 1965. Fluorometric determination of chlorophyll. *J. Con. perm. int. Explor*. 30, 3-15.

Koroleff, F., 1969. Direct determination of ammonia in natural waters as indophenol blue. *Int. Counc. Explor. Sea*, CM., 9, 19-22.

Landry, M.R. and Hassett, R.P., 1982. Estimating the grazing impact of marine microzooplankton. *Marine Biology*, 67, 283-288.

Law , C.S., Watson A.J., Liddicoat M.I., 1994. Automated vacuum analysis of sulphur hexafluoride in seawater: derivation of the atmospheric trend (1970-1993) and potential as a transient tracer. *Marine Chemistry*, 48, 57-69

Lorenzen, C.J., 1967. Determination of chlorophyll and phaeopigments: spectrophotometric equations. *Limnology and Oceanography*, 12.

Lorenzen, C.J. and Jeffrey, S.W., 1978. Determination of chlorophyll in seawater. *UNESCO Techn. Paper Mar Sci*, 35.

Miller, A.E.J, Mantoura, R.F.C., Preston, M.R. and Suzuki, Y, 1993. Preliminary study of DOC in the Tamar Estuary, UK, using UV-persulphate and HTCO techniques. *Marine Chemistry*., 41, 223-228.

Platt, T., Gallegos, C.L. and Harrison, W.G., 1980. Photoinhibition of photosynthesis in natural assemblages of marine phytoplankton. *J. Mar. Res.*, 38, 687-701.

Porter, K.G. and Feig, Y.S., 1980. Use of DAPI for identifying and counting aquatic microflora. *Limnology and Oceanography*, 25, 943-948.

Quinn, G.K., Charlston, R.J. and Zoller, W.H., 1987. *Tellus*, 39B, 413-425.

Quinn, G.K., Charlston, R.J. and Bates, T.S., 1988. *Nature*, 335, 120-121.

Rees, A.P., Owens, N.J.P. and Woodward, E.M.S. 1995. Phytoplankton nitrogen assimilation at low nutrient concentrations in the NW Mediterranean Sea. Water Pollution Research Report 32 in EROS 2000 ed J-M Martin and Barth, European Commission, 141-148.

Robinson, C., Williams, P.J.leB., 1991. Development and assessment of an analytical system for the accurate and continual measurement of total dissolved inorganic carbon. *Marine Chemistry* 34, 157-175

Roy, R.N., Roy, L.N., Vogel, K.M., Moore, C.P., Pearson, T., Good, C.E., Millero, F.J. and Cambell, 1993. Determination of the ionization constants of carbonic acid in seawater *Marine Chemistry* 44, 249-268.

Simon, M. and Azam, F., 1989. Protein content and protein synthesis rates of planktonic marine bacteria. *Marine Ecology Progress Series* 51, 201-213.

Strickland, J.D.H., Parsons, T.R. 1972. A practical handbook of seawater analysis. *Fish. Res. Bd. Can.*,.167-311.

Turner, S.M., Malin, G., Bagander, L.E. and Leck, C. 1990. Interlaboratory calibration and sample analysis of dimethylsulphide in water. *Marine Chemistry*. 29: 47-62.

Upstill-Goddard, R.C., Watson, A.J., Liss, P.S., Liddicoat, M.I. 1990. Gas transfer velocities in lakes measured with SF₆. *Tellus Ser B* 42, 364-377

Upstill-Goddard, R.C., Watson, A.J., Liddicoat, M.I., Wood, J., 1991. Sulphur hexafluoride and Helium-3 as seawater tracers: deployment techniques and continuous underway analysis of SF₆ by purge and trap ECGC. *Analytica Chimica Acta* 249, 555-562.

Weiss, R.F., 1974. Carbon dioxide in seawater: the solubility of a non-ideal gas. *Marine Chemistry* 2, 203-215.

Weiss, R.F., 1981. Determination of CO₂ and CH₄ by dual flame isolation chromatography and N₂O by electron capture chromatography. *Journal of Chromatographic Science*, 19, 611-616.

Wright, S.W., Jeffrey, S.W., Mantoura, R.F.C., Llewellyn, C.A., Bjornland, T., Repeta, D. and Welschmeyer, N., 1991. Improved HPLC method for the analysis of chlorophylls and carotenoids from marine phytoplankton. *Marine Ecology Progress Series*, 77, 183-196.